

Substrate Specificity Changes for Human Reticulocyte and Epithelial 15-Lipoxygenases Reveal Allosteric Product Regulation[†]

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ABSTRACT: Human reticulocyte 15-lipoxygenase (15-hLO-1) and epithelial 15-lipoxygenase (15-hLO-2) have been implicated in a number of human diseases, with differences in their substrate specificity potentially playing a central role. In this paper, we present a novel method for accurately measuring the substrate specificity of the two 15-hLO isozymes and demonstrate that both cholate and specific LO products affect substrate specificity. The linoleic acid (LA) product, 13-hydroperoxyoctadecadienoic acid (13-HPODE), changes the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio more than 5-fold for 15-hLO-1 and 3-fold for 15-hLO-2, while the arachidonic acid (AA) product, 12-(*S*)-hydroperoxyeicosatetraenoic acid (12-HPETE), affects only the ratio of 15-hLO-1 (more than 5-fold). In addition, the reduced products, 13-(*S*)-hydroxyoctadecadienoic acid (13-HODE) and 12-(*S*)-hydroxyeicosatetraenoic acid (12-HETE), also affect substrate specificity, indicating that iron oxidation is not responsible for the change in the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio. These results, coupled with the dependence of the 15-hLO-1 k_{cat}/K_m kinetic isotope effect ($^{\text{D}}k_{\text{cat}}/K_m$) on the presence of 12-HPETE and 12-HETE, indicate that the allosteric site, previously identified in 15-hLO-1 [Mogul, R., Johansen, E., and Holman, T. R. (1999) *Biochemistry* 39, 4801–4807], is responsible for the change in substrate specificity. The ability of LO products to regulate substrate specificity may be relevant with respect to cancer progression and warrants further investigation into the role of this product-feedback loop in the cell.

Human lipoxygenases (hLOs) make up a family of structurally related enzymes that catalyze the hydroperoxidation of polyunsaturated fatty acids using molecular oxygen (Scheme 1) (1). There are three main isozymes of pharmacological interest: 5-hLO, 12-hLO, and 15-hLO, which are named according to their positional specificity on arachidonic acid (AA),¹ producing their respective hydroperoxyeicosatetraenoic acid (HPETE) products. Each of these lipoxygenase isozymes plays a distinct biological role in human

disease. 5-hLO is implicated in asthma (2) and cancer (3, 4). 12-hLO is implicated in psoriasis (5) and cancer (4, 6, 7). 15-hLO is implicated in atherosclerosis (8) and cancer (4, 9).

Defining the exact role of LO in human disease is complicated by the incomplete understanding of three fundamental biochemical properties of hLO: substrate specificity, activation specificity, and allosteric regulation. With regard to substrate specificity, enzymes are typically highly specific and react with only one particular substrate, such as 5-hLO and platelet 12-hLO, which react with only AA. However, reticulocyte 15-hLO-1 and epithelial 15-hLO-2 react with both LA and AA, albeit with different efficiencies. 15-hLO-1 reacts preferentially toward LA (10), while 15-hLO-2 reacts preferentially toward AA (11), although there is a discrepancy in the literature (12) (*vide infra*). Interestingly, soybean lipoxygenase-1 (sLO-1), a plant homologue and model enzyme of 15-hLO-1, also reacts preferentially toward AA over LA, even though AA is not a native substrate in soybeans (13). The promiscuity in substrate preference for 15-hLO-1 and 15-hLO-2 is relevant to cellular biology since the products of LA, 13-hydroperoxyoctadecadienoic acid (13-HPODE), and AA, 15-HPETE, both affect cellular function. For example, in prostate cancer, 13-HPODE upregulates the MAP kinase signaling pathway, which causes cell proliferation and differentiation (14, 15), while 15-HPETE downregulates MAP kinase activity and lowers the degree of cell proliferation (16), implying a possible role for 15-hLO in cancer progression.

The fact that 15-hLO-1 and 15-hLO-2 react with both AA and LA is also interesting in a structural sense because it is

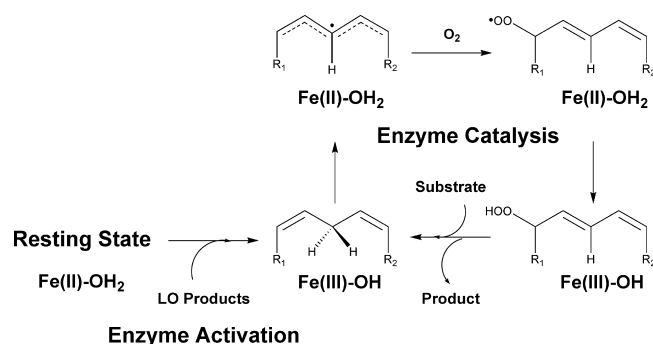
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¹ Abbreviations: LO, lipoxygenase; sLO-1, soybean lipoxygenase-1; 15-hLO-1, human reticulocyte 15-lipoxygenase-1; 15-hLO-2, human epithelial 15-lipoxygenase-2; 15-rLO, rabbit 15-lipoxygenase; AA, arachidonic acid; 15-HPETE, 15-(*S*)-hydroperoxyeicosatetraenoic acid (major oxidation product of AA with 15-hLO-1); 15-HETE, 15-(*S*)-hydroxyeicosatetraenoic acid; 12-HPETE, 12-(*S*)-hydroperoxyeicosatetraenoic acid (minor oxidation product of AA with 15-hLO-1); 12-HETE, 12-(*S*)-hydroxyeicosatetraenoic acid; LA, linoleic acid; 13-HPODE, 13-(*S*)-hydroperoxyoctadecadienoic acid (oxidation product of LA); 13-HODE, 13-(*S*)-hydroxyoctadecadienoic acid; *d*₃₁-LA, fully deuterated LA; perdeuterated 13-HPODE, fully deuterated 13-(*S*)-HPODE; perdeuterated 13-HODE, fully deuterated 13-(*S*)-HODE; k_{cat} , rate constant for product release; k_{cat}/K_m , rate constant for substrate capture; $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$, substrate capture ratio between AA and LA; $(k_{\text{cat}})^{\text{AA}}/(k_{\text{cat}})^{\text{LA}}$, product release ratio between AA and LA; $(k_{\text{cat}}/K_m)^{\text{Act}}/(k_{\text{cat}}/K_m)$, substrate capture ratio between product-activated enzyme and unactivated enzyme; $(k_{\text{cat}})^{\text{Act}}/(k_{\text{cat}})$, product release ratio between product-activated enzyme and unactivated enzyme; $^{\text{D}}k_{\text{cat}}/K_m$, kinetic isotope effect for k_{cat}/K_m .

Scheme 1



unclear exactly how either substrate binds in the active site or how substrate recognition occurs. It has previously been shown for 15-hLO-1 that there are three factors that participate in substrate recognition (17, 18). First, the depth of the active site (F352, I417, and M418) adjusts how far the methyl end of the substrate can enter, dictating which hydrogen atom is abstracted (regioselectivity). Second, R402 interacts with the carboxylic acid end of the substrate and affects the reactivity with AA and LA equally. This is an intriguing result since it indicates that R402 interacts with both AA and LA, even though the substrates are different in length by two carbons (~ 3 Å). This suggests that to accommodate the different lengths of AA and LA, either a global/local structural change in the enzyme, a conformational adjustment of the substrate, or both are required. Finally, F414 appears to facilitate π - π stacking with the ω -9 unsaturation, present in both AA and LA. In contrast, little is known about 15-hLO-2 substrate recognition; however, positional specificity has been studied, with D602, V603, and A416 of 15-hLO-2 affecting the distribution of AA products (19, 20). Notably, a comparison of 15-hLO-1 and 15-hLO-2 by sequence alignments indicates that only F352, discussed previously for 15-hLO-1, is conserved in 15-hLO-2 (F365), suggesting a possible difference in their molecular mechanism for substrate recognition.

The activation specificity of LO is another property which may potentially be relevant with respect to the role of LO in cellular function. LO isozymes are inactive in the as-isolated, ferrous state; however, 1 equiv of the hydroperoxide product converts the enzyme to the active, ferric state (Scheme 1) (1). Interestingly, the specific LO products which activate particular LO's are quite broad. 5-HPETE, 12-HPETE, 15-HPETE, and 13-HPODE all have been shown to activate 5-hLO (21, 22), while both 15-HPETE and 13-HPODE have been shown to activate 15-hLO-1 (17) and 15-hLO-2 (12). These data suggest that enzymatic products from one isozyme of LO can be used to activate another LO isozyme, potentially regulating each other's activity in the cell.

Allosteric regulation of protein activity is a common phenomenon in biology and is observed in various LOs. For example, ATP and Ca^{II} have been shown to bind to noncatalytic sites in 5-hLO, resulting in the activation of enzyme catalysis (23). For sLO-1, our laboratory has shown that an allosteric site exists, which can be specifically targeted by the allosteric inhibitor, oleyl sulfate (OS), to lower the enzymatic activity (24). In addition, we have observed an increase in the kinetic isotope effect of k_{cat}/K_m ($^Dk_{\text{cat}}/K_m$) for 15-hLO-1 with the addition of OS, further indicating the

presence of an allosteric site (24). These experimental results for 15-hLO-1, in conjunction with the other allosteric results for sLO-1 and 5-hLO, led us to surmise that the allosteric site in 15-hLO-1 may be biologically relevant and warranted further study.

This investigation utilizes a novel competitive substrate capture method to determine the substrate specificity of both 15-hLO-1 and 15-hLO-2. The use of this method resulted in the discovery that various LO products elicit changes in the substrate specificity for both 15-hLO isozymes and also change the $^Dk_{\text{cat}}/K_m$ for 15-hLO-1, indicating the presence of a product-activated, allosteric regulatory site for both 15-hLO isozymes.

MATERIALS AND METHODS

Materials. All commercial fatty acids (Sigma-Aldrich Chemical Co.) and LO products were repurified using a Higgins HAsil semipreparative (5 μm , 250 mm \times 10 mm) C-18 column. Solution A was 99.9% MeOH and 0.1% acetic acid; solution B was 99.9% H_2O and 0.1% acetic acid. An isocratic elution of 85% A and 15% B was used to purify all fatty acids, and then they were stored at -80°C for a maximum of 6 months. Perdeuterated LA (d_{31} -LA) (98% deuterated, Cambridge Isotope Laboratories) was purified as previously described (25). All other chemicals were reagent grade or better and were used without further purification.

Overexpression and Purification of sLO-1 and 15-Human Lipoxygenase. Overexpression and purification of sLO-1 followed a protocol outlined previously (26, 27). Human reticulocyte 15-lipoxygenase-1 (15-hLO-1) is an N-terminal, His₆-tagged protein which was expressed and purified as previously described (28). Human prostate epithelial 15-lipoxygenase-2 (15-hLO-2), without a His₆ tag, was expressed and purified as previously described for high-throughput screening (29). All enzymes were purified to >90% purity, as evaluated by SDS-PAGE analysis. The iron content of sLO-1 and the 15-hLO isozymes was determined with a Finnigan inductively coupled plasma mass spectrometer (ICP-MS), using cobalt-EDTA as an internal standard. Iron concentrations were compared to those of standardized iron solutions.

Determination of the Substrate Specificity of sLO-1 Using the Competitive Substrate Capture Method with AA and LA as Substrates. The competitive substrate capture method experiments were performed using reaction mixtures of AA and LA with a known molar ratio (1:1) and were initiated with sLO-1 (~ 1.0 nM, normalized to iron content) using the following buffer conditions: 100 mM borate, pH 9.2, and 22°C (no cholate added). The ratio of the simultaneous product formation (15-HPETE and 13-HPODE) by sLO-1 was determined at a total substrate concentration of 1 μM (substrate limiting conditions). The reaction was monitored at 234 nm with a Perkin-Elmer Lambda 40 instrument using an oversized reaction cuvette (path length of 5.4 cm) with a total volume of 60 mL and stopped with an acetic acid quench at $\sim 5\%$ total substrate consumption (~ 0.05 μM), similar to our previous experiments (25). The acidified reaction mixture was extracted with dichloromethane, evaporated to dryness under vacuum, reconstituted in 50 μL of MeOH, and injected onto a Phenomenex Luna (5 μm , 250 mm \times 4.6 mm) C-18 column. The elution protocol consisted

of 1 mL/min, an isocratic mobile phase of 54.9% ACN, 45% H₂O, and 0.1% acetic acid. The molar amount of 15-HPETE and 13-HPODE formation was equated to the corresponding peak areas determined by HPLC. The ratio of the peak areas was then used to determine the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio, modeled after our previous report (25).

Determination of the Substrate Specificity of 15-hLO Isozymes Using the Competitive Substrate Capture Method with an AA and LA Reaction Mixture. Competitive substrate capture experiments with the 15-hLO isozymes were performed as described above. Sodium cholate was not present since it hampers product extraction. Briefly, reactions were initiated by addition of 15-hLO-1 and 15-hLO-2 (~4 and ~20 nM, respectively, normalized to iron content) to an AA/LA mixture with a known molar ratio (1:1) (25 mM Hepes, pH 7.5, and 22 °C). The ratio of the simultaneous formation of 15-HPETE and 13-HPODE by the 15-hLO enzymes was determined as described above for sLO-1. However, it is important to note that 15-hLO-1 also produces 12-HPETE (13 ± 4% of total AA turnover), which could be observed by HPLC analysis and was considered a component of the total AA product concentration when the competitive substrate capture $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratios were calculated.

Determination of the Substrate Specificity of 15-hLO-1 and 15-hLO-2 Using Steady-State Kinetics with and without Sodium Cholate. Steady-state kinetics were determined with 15-hLO-1 and 15-hLO-2 in the presence and absence of sodium cholate (0.2%). Lipoygenase rates were determined by following the formation of the conjugated diene product at 234 nm ($\epsilon = 25000 \text{ M}^{-1} \text{ cm}^{-1}$) with a Perkin-Elmer Lambda 40 UV-vis spectrophotometer. All reactions were carried out in a volume of 2 mL and mixtures constantly stirred using a magnetic stir bar at room temperature (22 °C). Assays were carried out in 25 mM Hepes buffer (pH 7.5, 22 °C) with substrate concentrations ranging from 1 to 50 μM and from 1 to 30 μM with and without cholate, respectively. Assays performed in the presence of sodium cholate were initiated by addition of 15-hLO-1 (~90 nM for AA and ~20 nM for LA, with the enzyme concentration normalized to iron content) and 15-hLO-2 (~350 nM for both substrates, normalized to iron content). Assays performed without cholate were initiated by 15-hLO-1 (~25 nM for AA and ~15 nM for LA, normalized to iron content) and 15-hLO-2 (~300 nM for AA and ~750 nM for LA, normalized to iron content). Assays for each substrate were conducted concurrently to allow direct comparison of AA and LA kinetic data with and without detergent. Substrate concentrations were quantitatively determined by allowing the enzymatic reaction to go to completion. Kinetic data were obtained by recording initial enzymatic rates at each substrate concentration, which were then fitted to the Michaelis–Menten equation using KaleidaGraph (Synergy) to determine kinetic parameters k_{cat} and k_{cat}/K_m .

Determination of the Steady-State Substrate Specificity Using Activated 15-hLO-1 and 15-hLO-2 with Lipoygenase Products. To establish the concentration of product needed for the steady-state kinetic analysis, the purified products (13-HPODE, 15-HPETE, and 12-HPETE) were titrated to determine their effects on the rate of catalysis of 15-hLO isozymes under substrate limiting conditions (2.5 μM AA and LA). It was determined that the greatest effect on the rates, with the smallest amount of product, to avoid product

inhibition, was with 1 μM 15-hLO-1 and 10 μM 15-hLO-2. The difference in the amount of product required is due to the difference in protein concentration between the two isozymes. There was little change in the rates for both enzymes beyond these concentrations (data not shown). Steady-state kinetic data were obtained for both 15-hLO-1 and 15-hLO-2 in the presence of the three different LO products (13-HPODE, 15-HPETE, and 12-HPETE) to determine if activating the enzymes with products (i.e., oxidizing the active site ferrous ion and removing the lag phase) had an effect on substrate specificity. Steady-state kinetics for nonactivated enzyme were determined concurrently for direct comparison and to minimize error. Enzymatic assays were conducted using the same conditions as described for the steady-state kinetics without cholate present (25 mM Hepes, pH 7.5, 22 °C, no cholate); however, enzymes were incubated with LO products to facilitate enzyme–product interaction, followed by initiation of enzymatic assays with the addition of substrate. There was no observable difference in the initial rates with either addition of enzyme first and then initiation of the reaction with substrate or visa versa, as described above. The effects of product-activated enzyme on the kinetic parameters $[(k_{\text{cat}}/K_m)^{\text{Act}}/(k_{\text{cat}}/K_m)]$ and $(k_{\text{cat}})^{\text{Act}}/k_{\text{cat}}$ as well as the changes in the activated ratios $[(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ and $(k_{\text{cat}})^{\text{AA}}/(k_{\text{cat}})^{\text{LA}}$] were subsequently determined.

Determining the Effect of 12-HPETE and 12-HETE on the Substrate Specificity Using the Competitive Substrate Capture Method with 15-hLO-1. The competitive substrate capture reactions were carried out using the conditions described above (25 mM Hepes, pH 7.5, 22 °C) with the addition of 12-HPETE or 12-HETE (0.1, 0.5, 1.0, 2.0, and 5.0 μM). The enzymatic reactions were initiated by the addition of 1 μM substrate, after preincubation of product with 15-hLO-1 (~4 nM, normalized to iron content). The $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio was determined from the 15-HPETE:13-HPODE HPLC peak area ratio, as previously described. Since 12-HPETE and 12-HETE are the last products to elute from the C-18 column using the HPLC method, it was possible to add them without affecting the 13-HPODE and 15-HPETE peak areas, which is the key property that made this experiment possible. The amount of 12-HPETE generated by the isozyme was assumed to be constant (13% of the total AA turnover) because it could not be determined directly (the added 12-HPETE or 12-HETE saturated the signal) and because the 12-HPETE:15-HPETE peak ratio showed no significant change under all experimental conditions, including temperature or pH (data not shown). The reaction mixture was reduced with trimethylphosphite for HPLC analysis to ensure coelution of the 12-product added with the 12-product generated during the reaction. The determined amount of total product present in the final reaction mixture included 3.5 nM 12-HPETE, an approximation from the 5% enzymatic turnover.

Determining the Effect of Perdeuterated 13-HPODE and 13-HODE on Substrate Specificity Using the Competitive Substrate Capture Method with 15-hLO-1 and 15-hLO-2. The competitive substrate capture reactions were carried out using the conditions described above (25 mM Hepes, pH 7.5, 22 °C), with the addition of perdeuterated 13-HPODE or 13-HODE (0.1, 0.5, 1.0, 2.0, and 5.0 μM). The enzymatic reactions were initiated by the addition of 1 μM substrate,

after preincubation of product and enzyme, 15-hLO-1 and 15-hLO-2 (~4 and ~20 nM, respectively, normalized to iron content). The $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio was determined using the competitive substrate capture method as described above; however, the reduced products were quantitated with a Finnigan LTQ liquid chromatography–tandem mass spectrometry (LC–MS/MS) system. The titration of 13-HPODE or 13-HODE could not be performed directly since the added products would affect the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio calculation by modifying their HPLC peaks. However, perdeuterated 13-HPODE or 13-HODE can be distinguished from the other LO products using LC–MS ion peaks, which is the fundamental property that made this experiment possible. A Thermo Electron Corp. Aquasil (3 μm , 100 mm \times 2.1 mm) C-18 column was used to detect the reduced LO products with an elution protocol consisting of 0.2 mL/min and an isocratic mobile phase of 59.9% ACN, 40% H_2O , and 0.1% THF. The corresponding reduced product ion peak ratio was determined using negative ion MS/MS (collision energy of 35 eV), with the following masses: 15-HETE, parent at m/z 319 and fragments at m/z 175 and 219; 12-HETE, parent at m/z 319 and fragments at m/z 179 and 257; 13-HODE, parent at m/z 295 and fragments at m/z 183 and 251; and perdeuterated 13-HODE, parent at m/z 325 and fragments at m/z 213 and 281 (30). Using controls, we determined there was a difference in the MS detector response between the HODE and HETE products. A 1:1 13-HODE/15-HETE control mixture, with a concentration similar to that produced in our competitive experiment, yielded a ratio of 0.7 ± 0.05 , instead of a ratio of 1.0. There was a linear correlation between product concentration and ionization peak areas throughout the range of concentrations used in the competitive substrate capture experiments for both AA and LA products; therefore, all LC–MS data were systematically adjusted by 0.7. The amount of total product present in the final reaction includes 20 nM 13-HPODE, an approximation from the 5% enzymatic turnover.

Determining the Effects of 12-HPETE and 12-HETE on the Competitive Kinetic Isotope Effect of 15-hLO-1 with LA as the Substrate. The competitive KIE was performed as described above for the competitive substrate capture method (25); however, reactions were initiated by addition of a mixture of LA and d_{31} -LA, with a known molar ratio (1:2), to a preincubated mixture of 15-hLO-1 (~10 nM, normalized to iron content) and 12-HPETE or 12-HETE (5 μM). Enzymatic assays were performed under substrate limiting conditions (total substrate concentration of 5 μM), using buffer conditions described above (25 mM Hepes, pH 7.5). The addition of 12-HPETE or 12-HETE was performed at 22 °C, for comparison to previously published results (25). All extracted reaction mixtures were reduced with trimethylphosphite for HPLC analysis.

Construction of 15-hLO-1 and 15-hLO-2 Homology Models, Substrate Docking Studies, and Cavity Visualization. Open [Protein Data Bank (PDB) entry 1P0M, chain A, no inhibitor bound] and closed (PDB entry 1P0M, chain B, inhibitor bound) 15-hLO-1 and 15-hLO-2 homology models were built using the revised 2.4 Å resolution 15-rLO structure as a template (31). Coordinates from chain A (open) were translated to match chain B (closed) of 15-rLO prior to creation of the homology models, using Combinatorial Extension, which calculates structural alignment of two

models in three dimensions using the combinatorial extension method (32). 15-hLO-1 and 15-hLO-2 are 80 and 37% identical in sequence to 15-rLO, respectively. As previously described (33), the homology models were created using PLOP (distributed commercially as PRIME) which uses loop prediction (34), side chain prediction (35, 36), and energy minimization to construct the models. The homology model construction is based on alignments obtained from NCBI-BLAST (blastp) (37). Briefly, primary structural modifications that must be made during the initial construction are the closing of chain breaks associated with gaps in the sequence alignment and side chain optimization for all residues not identical between the target and template sequences. Chain break closure is accomplished using an iterative application of a loop prediction algorithm as previously described in detail (34). After chain breaks have been closed, side chain optimization (35, 36) and complete energy minimization are performed on all portions of the protein whose coordinates either were not taken from the template at all or were modified during the building procedure. The homology modeling procedure uses the OPLS all-atom force field (35, 38, 39) and the Generalized Born solvent model (40, 41) for choosing low-energy structures. As in previous studies (33), flexible ligand docking of the substrates was performed using Glide (Schrödinger, Inc.) (42, 43), which uses a modified version of the Chemscore energy function to score the protein–ligand interactions (44). The optimized force field reduces the net ionic charge on formally charged groups by ~50% to make the gas-phase Coulombic interaction energy a better predictor of binding. The charge–charge interaction is not a qualification, but a contributor to docked poses. Molecules were docked using either the standard precision mode (SP) or the extra precision mode (XP), which uses a more optimized scoring function as well as a more extensive search of ligand conformations than the SP mode. Clefts and interior surfaces for 15-hLO-1 and 15-hLO-2 homology models were calculated using Hollow (45). Hollow generates volume-filling atoms which can be visualized as surfaces. A surface probe of 0.8 Å was used for the calculations, and [−57, 170, 43] and [−48, 161, 31] were chosen as the limits of the grid, which allow visualization (PYMOL) of only the active site cavity (46).

RESULTS

Overexpression and Purification of sLO-1 and 15-Human Lipoxygenase. Protein expression yields were similar to previously published results (26, 27, 47). The metal content was determined to be 90 ± 5 , 33 ± 2 , and $43 \pm 2\%$ for sLO-1, 15-hLO-1, and 15-hLO-2, respectively. All kinetic data were normalized to metal content.

Determination of the Substrate Specificity of sLO-1 Using the Competitive Substrate Capture Method with AA and LA as Substrates. The $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio for sLO-1 using the competitive substrate capture method (100 mM borate, pH 9.2, 22 °C) was determined to be 1.8 ± 0.2 . This is in agreement with previous steady-state kinetic data (1.8 ± 0.5 ; 100 mM borate, pH 9.2, 22 °C) (13) and confirms the validity of the competitive substrate capture method.

Determination of the Substrate Specificity of 15-hLO-1 and 15-hLO-2 Using the Competitive Substrate Capture Method

Table 1: Comparison of Steady-State Kinetic Parameters for 15-hLO Isozymes^a

	AA		LA		AA/LA	
	k_{cat}/K_m ($\mu\text{M}^{-1} \text{s}^{-1}$)	k_{cat} (s^{-1})	k_{cat}/K_m ($\mu\text{M}^{-1} \text{s}^{-1}$)	k_{cat} (s^{-1})	k_{cat}/K_m	k_{cat}
15-hLO-1						
cholate	0.45 ± 0.04	1.7 ± 0.08	1.8 ± 0.2	8.5 ± 0.6	0.25 ± 0.05	0.20 ± 0.02
no detergent	2.0 ± 0.2	5.3 ± 0.4	2.5 ± 0.2	7.8 ± 0.4	0.80 ± 0.1	0.68 ± 0.09
15-hLO-2						
cholate	0.026 ± 0.002	0.33 ± 0.02	0.029 ± 0.002	0.31 ± 0.02	0.90 ± 0.1	1.1 ± 0.1
no detergent	0.10 ± 0.01	0.74 ± 0.03	0.013 ± 0.001	0.14 ± 0.01	8.0 ± 1.0	5.3 ± 0.5

^a Enzymatic assays were performed in 25 mM Hepes (pH 7.5) at 22 °C.

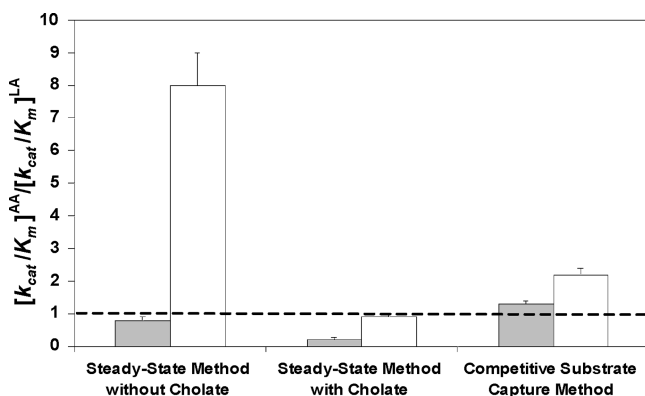


FIGURE 1: Comparison of $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratios from steady-state kinetics and the competitive substrate capture method for 15-hLO-1 and 15-hLO-2 (gray and white, respectively). The dashed line represents no substrate specificity (ratio of 1). A ratio of >1 represents AA specificity; a ratio of <1 represents LA specificity. Enzymatic assays were performed in 25 mM Hepes at pH 7.5 and 22 °C.

with an AA/LA Reaction Mixture. The competitive substrate capture experiment (25 mM Hepes, pH 7.5, 22 °C) was utilized as an alternative method to determine the substrate specificity for both 15-hLO isozymes. There was no change in the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio from 5 to 10% total substrate turnover and at total substrate mixture concentrations between 1 and 5 μM . Therefore, experiments were conducted with an initial substrate concentration of 1 μM ($\sim 5\%$ total substrate turnover and a 1:1 AA/LA mixture), with no inactivation of either 15-hLO isozyme being observed. The $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio was thus determined to be 1.3 ± 0.1 and 2.2 ± 0.2 for 15-hLO-1 and 15-hLO-2, respectively.

Determination of the Substrate Specificity of 15-hLO-1 and 15-hLO-2 Using Steady-State Kinetics with and without Sodium Cholate. The kinetic data with and without sodium cholate (25 mM Hepes, pH 7.5, 22 °C) for 15-hLO-1 and 15-hLO-2 are listed in Table 1 (standardized to iron content) and compared to those from the competitive substrate capture method (Figure 1).

Unlike the sLO-1 data, the ratios for the 15-hLO isozymes are markedly different from that of the competitive method (Figure 1). Briefly, cholate reduced both the k_{cat}/K_m and k_{cat} of 15-hLO-1 and 15-hLO-2 with AA as the substrate; however, cholate increased both the k_{cat}/K_m and k_{cat} of 15-hLO-2 with LA as the substrate. Consequently, the AA:LA ratios were reduced for both the kinetic parameters and both the isozymes with cholate present. The 15-hLO-1 substrate specificity ratio is in excellent agreement with the previous steady-state result (0.25 and 0.20, respectively) (10); how-

Table 2: Comparison of the Steady-State Product-Activated Kinetic Parameter $(k_{\text{cat}}/K_m)^{\text{Act}}/(k_{\text{cat}}/K_m)$ and Activated $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ^a

product	$(k_{\text{cat}}/K_m)^{\text{Act}}/(k_{\text{cat}}/K_m)$		$(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$
	AA	LA	AA/LA
15-hLO-1			
none	—	—	0.8 ± 0.1
13-HPODE	1.8 ± 0.4	1.2 ± 0.4	1.6 ± 0.5
15-HPETE	1.2 ± 0.4	1.1 ± 0.4	1.0 ± 0.4
12-HPETE	2.7 ± 0.5	1.0 ± 0.1	2.0 ± 0.5
15-hLO-2			
none	—	—	8.0 ± 1.0
13-HPODE	0.8 ± 0.2	1.7 ± 0.6	3.9 ± 0.9
15-HPETE	1.4 ± 0.4	1.3 ± 0.3	8.0 ± 3.0
12-HPETE	1.2 ± 0.4	1.4 ± 0.4	7.0 ± 2.0

^a Enzymatic assays were performed in 25 mM Hepes (pH 7.5) at 22 °C, with 1 and 10 μM product added to 15-hLO-1 and 15-hLO-2, respectively.

ever, our $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio for 15-hLO-2, with cholate present, is 2-fold greater than that of Kilty et al. (0.9 ± 0.1 and 0.5 , respectively) (12). Considering that our 15-hLO-1 data are consistent with the literature, we propose that the difference between our 15-hLO-2 data and those of Kilty et al. is due to experimental differences. Kilty et al. added 13-HPODE (2 μM) to product-activate the enzyme (12), which we currently show affects the LO kinetics (*vide infra*), and utilized exceedingly high substrate concentrations ($>150 \mu\text{M}$), which may lead to formation of micelles at pH 7.4.

Determination of the Steady-State Substrate Specificity Using Activated 15-hLO-1 and 15-hLO-2 with Lipoygenase Products. The discrepancy between the steady-state kinetic data and the competitive substrate capture method data was investigated using three LO products (13-HPODE, 12-HPETE, and 15-HPETE) to activate the 15-hLO isozymes. The data are presented in two forms: the ratio of kinetic parameters from product-activated to unactivated [$(k_{\text{cat}}/K_m)^{\text{Act}}/(k_{\text{cat}}/K_m)$ and $(k_{\text{cat}})^{\text{Act}}/k_{\text{cat}}$ (Tables 2 and 3, respectively)] and the ratio of kinetic parameters from the product-activated enzyme with AA and LA as substrates [product-activated $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ and product-activated $(k_{\text{cat}})^{\text{AA}}/(k_{\text{cat}})^{\text{LA}}$ (Tables 2 and 3, respectively)].

The k_{cat}/K_m values increased with the addition of most of the three products; however, only certain LO products elicited a change in the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio (Table 2). For 15-hLO-1, 13-HPODE increased the steady-state k_{cat}/K_m for both LA and AA, but the increase for AA was greater, resulting in a net increase in the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio toward AA. The LO product, 12-HPETE, however, did not affect LA kinetics but did increase the k_{cat}/K_m for AA,

Table 3: Comparison of Steady-State Product-Activated Kinetic Parameter $(k_{\text{cat}})^{\text{Act}}/k_{\text{cat}}$ and Activated $(k_{\text{cat}})^{\text{AA}}/(k_{\text{cat}})^{\text{LA}}$ ^a

product	$(k_{\text{cat}})^{\text{Act}}/k_{\text{cat}}$		$k_{\text{cat}}/k_{\text{cat}}$
	AA	LA	AA/LA
15-hLO-1			
none	—	—	0.68 ± 0.09
13-HPODE	0.9 ± 0.1	0.9 ± 0.1	0.8 ± 0.2
15-HPETE	1.0 ± 0.1	0.9 ± 0.1	0.7 ± 0.2
12-HPETE	1.1 ± 0.2	0.7 ± 0.1	0.9 ± 0.2
15-hLO-2			
none	—	—	5.3 ± 0.5
13-HPODE	0.8 ± 0.1	1.0 ± 0.1	5.3 ± 0.4
15-HPETE	0.9 ± 0.1	0.9 ± 0.1	4.6 ± 0.9
12-HPETE	0.9 ± 0.1	0.8 ± 0.2	4.6 ± 0.9

^a Enzymatic assays were performed in 25 mM Hepes (pH 7.5) at 22 °C, with 1 and 10 μM product added to 15-hLO-1 and 15-hLO-2, respectively.

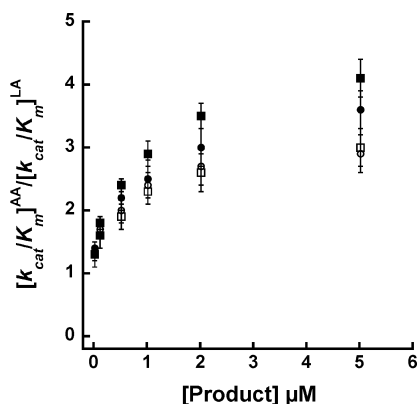


FIGURE 2: Competitive substrate capture method saturation curves of $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ vs product concentration (micromolar) for 15-hLO-1. Data demonstrate similar effects on substrate specificity for the AA products, 12-HPETE (■) and 12-HETE (□), and for the perdeuterated LA products, 13-HPODE (●) and 13-HODE (○). Enzymatic assays were performed in 25 mM Hepes at pH 7.5 and 22 °C.

resulting in an increased $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio toward AA. For 15-hLO-2, 13-HPODE increased the k_{cat}/K_m for LA but decreased the k_{cat}/K_m for AA, which resulted in a net decrease in the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio toward AA. However, 15-hLO-2 displayed little change in the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio with the addition of either HPETE product. The k_{cat} ratios were only slightly changed with the addition of LO products for both 15-hLO-1 and 15-hLO-2 (Table 3), indicating that the effect of product addition is principally targeting substrate capture, a common feature in allosteric regulation (48).

Determining the Effect of 12-HPETE and 12-HETE on Substrate Specificity Using the Competitive Substrate Capture Method with 15-hLO-1. The addition of 12-HPETE or 12-HETE produced a hyperbolic saturating effect on the substrate specificity for 15-hLO-1, demonstrating $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratios ranging from 1.3 ± 0.2 to 4.1 ± 0.3 and from 1.3 ± 0.2 to 3.0 ± 0.3, respectively (Figure 2). The data were fit to a saturation curve by plotting $\Delta(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ versus the micromolar concentration of 12-HPETE or 12-HETE (data not shown), and the half-saturation points were determined to be 1.0 ± 0.2 and 1.1 ± 0.1 μM, with the maximum effect being 4.6 ± 0.2 and 3.5 ± 0.2, at saturating levels of 12-HPETE and 12-HETE, respectively.

Determining the Effect of Perdeuterated 13-HPODE and 13-HODE on Substrate Specificity Using the Competitive Substrate Capture Method with 15-hLO-1 and 15-hLO-2. The addition of perdeuterated 13-HPODE or 13-HODE products had a hyperbolic saturating effect on the substrate specificity for 15-hLO-1, demonstrating $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratios ranging from 1.4 ± 0.2 to 3.6 ± 0.3 and from 1.4 ± 0.2 to 2.9 ± 0.3, respectively (Figure 2). The data were fit to a saturation curve by plotting $\Delta(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ versus micromolar 13-HPODE or 13-HODE concentration (data not shown), and the half-saturation points were determined to be 1.2 ± 0.2 and 0.8 ± 0.1 μM, with the maximum effect being 4.1 ± 0.3 and 3.2 ± 0.2, respectively, under saturating conditions. For 15-hLO-2, however, the perdeuterated 13-HPODE had no effect on the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio, with an average ratio of 2.3 ± 0.3, at all concentrations of perdeuterated 13-HPODE added (data not shown). It should be noted that the 15-hLO-2 competitive ratio, with 5 μM 13-HPODE added, does not match the steady-state ratio, with 10 μM 13-HPODE added (2.2 ± 0.2 and 3.9 ± 0.9, respectively). This discrepancy is most likely because the concentration of the 15-hLO-2 enzyme that is necessary to detect the steady-state initial rates is comparable to the substrate concentration. Consequently, the enzyme is no longer under steady-state approximation conditions, which leads to a large error in the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio.

Determining the Effects of 12-HPETE and 12-HETE on the Competitive Kinetic Isotope Effect of 15-hLO-1 with LA as the Substrate. Previous mechanistic characterization, using the competitive KIE experiment, demonstrated that 15-hLO-1 displayed a temperature-dependent $^Dk_{\text{cat}}/K_m$, with LA as the substrate. The addition of 12-HPETE or 12-HETE (5 μM) saturated the $^Dk_{\text{cat}}/K_m$ by increasing it from 24 ± 4 to 48 ± 10 and 44 ± 8 at 22 °C, respectively. These results are comparable to the previously published $^Dk_{\text{cat}}/K_m$ results at high temperatures (25) and similar to the effect of the allosteric inhibitor, oleyl sulfate (OS), on the $^Dk_{\text{cat}}/K_m$ of 15-hLO-1 (24), demonstrating allosteric binding of LO products to 15-hLO-1.

Construction of 15-hLO-1 and 15-hLO-2 Homology Models, Substrate Docking Studies, and Cavity Visualization. Homology models, docking simulations, and molecular visualization were carried out as described in Materials and Methods. AA and LA were successfully docked in the closed 15-hLO-1 and closed 15-hLO-2 homology models using the SP mode; however, the substrates did not dock to the open models of either isozyme. The modes of binding for AA and LA with the closed 15-hLO-1 model were consistent with mutagenesis studies (17, 18). The carboxylate end of the fatty acid formed a salt bridge with R402, which has no interaction with other residues prior to substrate binding in either the 15-rLO structure or our 15-hLO-1 model. The methyl end has hydrophobic interactions with I417 and M418, and ω-6 and ω-9 are in the proximity of F414 (Figure 3). The LA and AA poses are similar to those in previous docking studies with the charge interaction at R402 (R707 in sLO) and the hydrophobic interaction at the methyl terminus (17, 49, 50). Furthermore, it has been previously demonstrated through mutagenesis that F352 is a primary determinant for the regiospecificity of mammalian 15-LOs (18), which is also found at the methyl end of the homology model active site of 15-hLO-1 (residue not shown in Figure 3 for clarity).

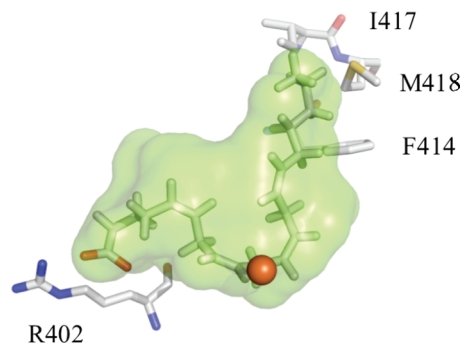


FIGURE 3: Depiction of the closed 15-hLO-1 homology model, active site cavity (green), with a docked AA molecule. Selected residues, observed to be catalytically important, are displayed (R402, F414, I417, and M418, with the red sphere as the iron).

Table 4: Changes in the Distances of the α -Carbon and Moiety for Residues within 6 Å of Docked AA to 15-hLO-1^a

residue	α -carbon	Δ distance (Å)	residue moiety	moiety Δ distance (Å)
K170		5.6	C(NH ₂)	3.4
R171		6.2	C(NH ₂) ₂	16
V172		9.4	C(CH ₃)	12
D173		9.4	C(CO ₂)	10
F174		10	ring C1	3.4
E175		10	C(CO ₂)	11
S177		12	COH	13
L178		12	C(CH ₃) ₂	12
R402		0.6	C(NH ₂) ₂	0.8
W594		2.3	ring C1	3.2
Q595		4.0	C(C=O)NH ₂	4.8
L596		5.6	C(CH ₃) ₂	5.6
G597		5.8	CH ₂	5.8
R598		5.0	C(NH ₂) ₂	10
Q600		2.2	C(C=O)NH ₂	3.5

^a Changes in distances were determined from the open vs closed forms of 15-hLO-1 homology models.

These observations, along with our previous use of a 15-hLO-1 model for LO inhibitor discovery (33), lead to this docking model having credibility.

The fatty acid docking to the closed 15-hLO-2 model, however, is less straightforward. The majority of substrate docking poses of both AA and LA were in the reverse orientation compared to that seen for 15-hLO-1, with S430 forming a hydrogen bond with the carboxylate of the substrate. This is most likely due to the fact that there is no positively charged residue in the 15-hLO-2 active site, such as R402 in 15-hLO-1, which is the primary determinate for substrate orientation (17). In fact, none of the residues found by mutagenesis to be important for substrate recognition in 15-hLO-1 are conserved in 15-hLO-2, except for F365 (F352 in 15-hLO-1). The substrates were subsequently docked in the more accurate XP mode, due to this unusual orientation of the substrate in the closed 15-hLO-2 model, and a salt bridge between the carboxylate of AA and R429 was observed (data not shown). However, the long distance between R429 and the active site (>17 Å) does not allow proper positioning of the substrate for hydrogen atom abstraction, with the methyl end of the substrate being near the iron, indicating this is an unlikely substrate orientation for catalysis.

Representations of the active site cavity for both 15-hLO isozymes were created using Hollow, with the active site dimensions being constrained to depict only the substrate-binding cavity. A similar approach was used previously with

various plant LO isozymes, to demonstrate differences in their active sites (51). As observed in Figure 4, the superposition of the open (red) and closed (green) active site cavities of 15-hLO-1 clearly shows a marked change in the 15-hLO-1 structure, similar to that seen for the 15-rLO structure (31).

The closed active site demonstrates the classic “U” or “boot” shape of the cavity, allowing the substrate to bend around the active site iron. The open form is markedly different, with an expansion of the cavity near the substrate entrance by the α -2 helix and a contraction of the cavity in the interior, which appears to hamper our ability to dock substrates to the open form. To improve our understanding of this movement, a 6 Å sphere was created around the docked AA to 15-hLO-1 and the active site amino acids with root-mean-square deviation (rmsd) values greater than 2.0 Å were recorded (Table 4), with the majority being near the carboxylate end of the bound substrate. The flexibility of this region of the active site may have relevance with respect to the substrate specificity and recognition, since the carboxylate end of the substrate is where AA and LA have the greatest difference.

When the closed cavities of 15-hLO-1 and 15-hLO-2 are compared, they display similar shapes and volumes, with the cavity of 15-hLO-2 being slightly larger (Figure 5). However, there are no positively charged residues, within a reasonable vicinity of the 15-hLO-2 active site, to facilitate binding of the substrate’s carboxylate moiety, as seen in 15-hLO-1. This lack of similarity between the electrostatic surfaces of the two cavities, as well as a lack of literature regarding known residues involved in substrate recognition for 15-hLO-2, makes it difficult to assess the catalytically productive substrate orientation and the corresponding structural constraints for 15-hLO-2.

DISCUSSION

Our laboratory has had a continued interest in the molecular mechanism of substrate recognition and allostery for lipoxygenase, and whether there are similarities between all lipoxygenases, be they from the plant or mammalian kingdom. One of the more thoroughly investigated LO isozymes, with respect to substrate recognition, is 15-hLO-1. For this reason, it was especially interesting that the substrate specificity of 15-hLO-2 was found to be different from that of 15-hLO-1 (11, 12, 17), suggesting a possible difference in their substrate recognition mechanisms. We therefore decided to investigate the substrate specificity of both 15-hLO-1 and 15-hLO-2 in further detail; however, the facts that the catalytic activity of 15-hLO-2 is exceedingly low against LA and that the addition of product is required to remove its long lag phase, prompted us to develop a novel method for measuring the substrate specificity $[(k_{cat}/K_m)^{AA}/(k_{cat}/K_m)^{LA}]$. This method is based on competitive substrate capture and measures the ratio of product turnover in the presence of both substrates, which is directly related to the $(k_{cat}/K_m)^{AA}/(k_{cat}/K_m)^{LA}$ ratio. In addition, the competitive experiment is performed at low substrate concentrations (1 μ M total substrate concentration) and is therefore a better mimic of the cellular conditions compared to steady-state methods.

The reliability of the novel competitive substrate capture method was confirmed using the model lipoxygenase, sLO-

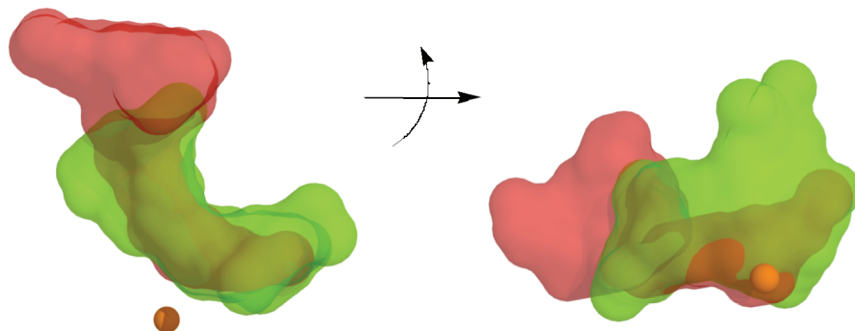


FIGURE 4: 15-hLO-1 homology model cavities from the open form (red) and closed form (green) superimposed and viewed from two perspectives, with an approximate 90° rotation about the horizontal axis. Note that the closed 15-hLO-1 cavity perspective, on the right side, is the same as that in Figure 3. The ferrous ion is represented as a sphere, and the cavities were created using Hollow.

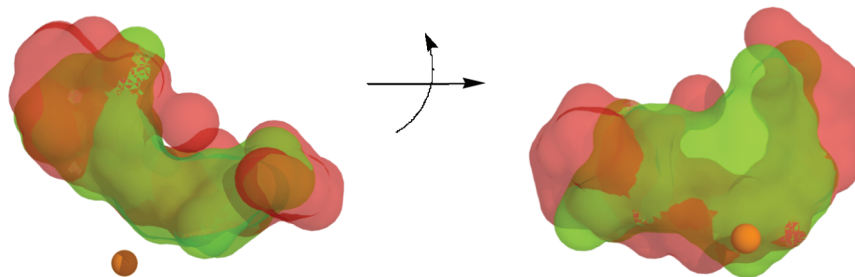


FIGURE 5: Homology model cavities for the closed forms of 15-hLO-1 (green) and 15-hLO-2 (red) superimposed and viewed from two perspectives, with an approximate 90° rotation about the horizontal axis. Note that the closed 15-hLO-1 cavity perspective, on the right side, is the same as that in Figures 3 and 4. The ferrous ion is represented as a sphere, and the cavities were created using Hollow.

1, which demonstrated a $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio similar to that of previously published steady-state results (1.8 ± 0.2 and 1.8 ± 0.5 , respectively) (13). The competitive substrate capture data for the two 15-hLO isozymes, however, did not match their published steady-state kinetic data. The competitive substrate capture method demonstrated that both 15-hLO-1 and 15-hLO-2 exhibit specificity toward AA [$(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}} = 1.3 \pm 0.1$ and 2.2 ± 0.2 , respectively], which was significantly different from the LA preference previously determined using steady-state kinetics [$(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}} = 0.2$ and 0.5 , respectively] (10, 12). Considering that the sLO-1 competitive results were consistent with the steady-state results, the discrepancy of the 15-hLO data was suspected to be due to experimental differences between the two methods: the presence of cholate and the number of substrate/product pairs (e.g., LA/13-HPODE or AA/15-HPETE) in the reaction mixture.

With cholate present (0.2% or 4.6 mM), we observed that the steady-state $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio for both 15-hLO-1 and 15-hLO-2 decreased (Table 1 and Figure 1). These data indicate that cholate is not simply a substrate solubilizer but that it also affects enzymatic activity. The consequence of detergents has already been investigated with respect to plant lipoxygenases (52), but this is the first example of a direct comparison of the effect of cholate on both the 15-hLO-1 and 15-hLO-2 kinetic parameters. Nevertheless, it is not straightforward to assign the role of cholate in LO catalysis. Cholate acts as an inhibitor toward 15-hLO-1 and 15-hLO-2 with AA as the substrate, but as an activator toward 15-hLO-2 with LA as the substrate (Table 1), effectively changing the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio for both 15-hLO-1 (from 0.8 ± 0.1 without cholate to 0.25 ± 0.05 with cholate) and 15-hLO-2 (from 8.0 ± 1.0 without cholate to 0.9 ± 0.1 with cholate) (Table 1 and Figure 1). It should be noted that even though the concentration of cholate is below its critical

micelle concentration, suggesting an individual molecular interaction, it is still high enough (4.6 mM) to indicate a weak, nonspecific interaction. These data offer an explanation for the discrepancy in the literature for the substrate specificity of 15-hLO-2 (11, 12), since experimental conditions, such as the presence of cholate or LO products (*vide infra*), have large effects on substrate specificity.

The dual nature of cholate (activator and inhibitor), along with the lack of correlation between the competitive and the steady-state results, led us to suspect that an allosteric site, as already observed for 15-hLO-1 (24), was affecting the substrate specificity of the 15-hLO isozymes. These facts compelled us to consider the second difference between the competitive method and the steady-state method, the presence of two substrate/product pairs. To determine if LO products affect substrate specificity, the steady-state kinetic method was performed with the addition of purified hydroperoxide products (13-HPODE, 12-HPETE, and 15-HPETE), without cholate present. The first observation was that all three products removed the kinetic lag phase for both 15-hLO-1 and 15-hLO-2, indicating that all three products oxidize the active site iron and are pseudoperoxidase substrates. These results are consistent with previous activation data for 5-hLO, 15-hLO-1, and 15-hLO-2 (12, 17, 21, 22), in which a variety of LO products are pseudoperoxidase substrates for these hLO isozymes.

The more significant result from this experiment was that certain LO products changed the substrate specificity of the 15-hLO isozymes (Table 2). The LA product, 13-HPODE, increased the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio toward AA for 15-hLO-1 but decreased the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio toward AA for 15-hLO-2. The AA product, 12-HPETE, also increased the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio toward AA for 15-hLO-1 but did not affect the ratio for 15-hLO-2. Furthermore, little effect was seen with the addition of 15-HPETE to either

15-hLO enzyme. These data indicate that alteration of the substrate specificity is a highly selective event, with discrimination between structurally similar LO products, such as 12-HPETE and 15-HPETE. In addition, the LO products have opposite effects depending on the 15-hLO isozyme, with the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio increasing for 15-hLO-1 yet decreasing for 15-hLO-2 with the addition of 13-HPODE. These combined results suggest that these specific LO products are binding to an allosteric site, which directly affects substrate specificity.

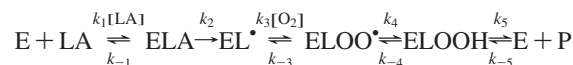
We subsequently utilized our novel competitive method for confirmation of these unusual results. The competitive substrate capture method, in which 12-HPETE or perdeuterated 13-HPODE was titrated to 15-hLO-1, demonstrated saturating effects on the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio and revealed a half-saturation point of 1.0 ± 0.1 or 1.2 ± 0.2 μM and a maximum effect on the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio of 4.6 ± 0.2 or 4.1 ± 0.2 , respectively (Figure 2). This is a greater than 5-fold change in the substrate specificity ratio with the addition of either 12-HPETE or 13-HPODE and confirms that addition of the LO product directly affects the substrate specificity for 15-hLO-1. However, when perdeuterated 13-HPODE was added to 15-hLO-2 using the competitive substrate capture method, no effect was seen. This is an intriguing result since the substrate specificity for the competitive method is clearly different from that of the steady-state method (2.2 ± 0.2 and 8.0 ± 1.0 , respectively) (Figure 1). We speculate that the small amount of 13-HPODE generated from LA being present in the competitive method may be sufficient to saturate the effect on the 15-hLO-2 substrate specificity. This would imply that 13-HPODE has a higher affinity for 15-hLO-2 than that seen for 15-hLO-1. We are currently investigating the affinity of the LO products to the allosteric site for both 15-hLO-1 and 15-hLO-2 directly, but due to the low percentage of iron loading and multiple possible binding sites for the LO products, catalytic, activation, and allosteric, this has proven to be more difficult than we first perceived. Finally, the LO products do not affect either the regioselectivity (12-HPETE vs 15-HPETE) (53), indicating that the allosteric effect is very specific to substrate specificity.

These results are strong evidence that both 15-hLO-1 and 15-hLO-2 have allosteric sites which bind LO products; however, there is a possibility that the substrate specificity change is linked to the ability of the LO products to oxidize the active site iron. It has been previously shown that under certain conditions, the radical intermediate can be lost during catalysis, leaving the inactive, ferrous species (54). If the rate of loss of the intermediate is different for LA than AA, and the rate of reactivation by the various products is different as well, then it is conceivable that the change in substrate specificity could be coupled to iron oxidation, not to allosteric effector properties of the LO products. To rule out this possibility, the aforementioned competitive method titration experiments were repeated using the reduced products, perdeuterated 13-HODE and 12-HETE. As shown in Figure 2, the reduced products have effects on the substrate specificity similar to those of the hydroperoxide products, in terms of both half-saturation and maximal effect, which clearly indicate that the substrate specificity effect is unrelated to oxidation of the active site iron. Another possible explanation for the substrate specificity effect is that the

products induce substrate, micelle formation, which subsequently affects the substrate specificity. This is unlikely since LO products have been shown to disrupt lipid bilayers (55), and the substrate specificity effect is seen with only a limited number of LO products and not the entire family of hydroperoxides. In total, these data strongly support the hypothesis that the change in substrate specificity with the addition of LO products is due to an allosteric interaction and not due to active site iron oxidation or product aggregation.

Allosteric regulation is a common feature in biology that can either activate or inhibit enzymatic activity. In the case of 5-hLO, ATP and Ca^{II} bind to allosteric effector sites and activate enzyme catalysis (23). Allosterism has also been observed with sLO-1, in which oleyl sulfate (OS) partially inhibits the enzyme activity but does not interfere with enzyme activation (24, 25, 27). In addition, allosteric binding of OS increases the observed $^Dk_{\text{cat}}/K_m$ for both sLO-1 and 15-hLO-1. This change in $^Dk_{\text{cat}}/K_m$ can be explained by examining the proposed mechanism of LO, minimally described using Scheme 2 (24, 25, 27).

Scheme 2



$$\text{KIE} = {}^Dk_{\text{cat}}/K_m = (k_{\text{cat}}/K_m)^{\text{H}}/(k_{\text{cat}}/K_m)^{\text{D}} = (k_2^{\text{H}}/k_2^{\text{D}} + k_2^{\text{H}}/k_{-1}^{\text{H}})/(1 + k_2^{\text{H}}/k_{-1}^{\text{H}}) \quad (1)$$

According to Scheme 2, $^Dk_{\text{cat}}/K_m$ would be described by eq 1, where substrate release (k_{-1}) and C–H bond cleavage (k_2) are the primary determinants for $^Dk_{\text{cat}}/K_m$ (this assumes $k_2 = k_{\text{cat}}$ and the multiple steps observed at low temperature are included in k_2). The $^Dk_{\text{cat}}/K_m$ increases to a maximum of $k_2^{\text{H}}/k_2^{\text{D}}$ when commitment (k_2/k_{-1}) is small and decreases to approaching 1 when commitment is large, assuming that the intrinsic $k_2^{\text{H}}/k_2^{\text{D}}$ remains unchanged. With respect to 12-HPETE and 12-HETE, both products have an effect on the $^Dk_{\text{cat}}/K_m$ of 15-hLO-1 similar to that of OS (24), increasing the $^Dk_{\text{cat}}/K_m$ at 22 °C from 24 ± 4 to 48 ± 10 and 44 ± 8 , respectively. These $^Dk_{\text{cat}}/K_m$ data indicate that the LO products lower commitment (k_2/k_{-1}), which is supported by the steady-state data where the addition of 12-HPETE lowers the k_{cat} (i.e., k_2) of 15-hLO-1 (Table 2). These $^Dk_{\text{cat}}/K_m$ data also support the substrate specificity data, which indicate that specific LO products bind as allosteric effectors to the 15-hLO isozymes and change the enzymatic mechanism; however, it is unclear what conformational change occurs to elicit such a change. The allosteric regulation of substrate specificity by an enzyme's own product is a relatively rare occurrence in biology, with ribonucleotide reductase (RNR) being one of the few enzymes in the literature to regulate its substrate specificity via binding its own products as allosteric effectors (56). The mechanism for this effect in RNR involves the product binding close to the active site and modulating the conformation of a protein loop, which in turn affects the active site and substrate specificity (57).

In an attempt to hypothesize possible structural changes in the active site which could change substrate specificity, we have utilized molecular modeling to compare the active sites of both 15-hLO-1 and 15-hLO-2. As mentioned in the

introductory section, there are specific residues in 15-hLO-1 that play a role in substrate recognition which are not present in the 15-hLO-2 enzyme. Therefore, we have generated computer models for both isozymes based on the 15-rLO crystal structures (both closed with inhibitor bound and open without inhibitor) (31) and docked both AA and LA into their active sites to improve our understanding of the binding modes of these substrates. The first observation is that both LA and AA were successfully docked into the closed active site of 15-hLO-1, with clear interactions with R402, F352, F414, I417, and M418, corroborating the results from previous mutagenesis studies (Figure 3). Considering that the methyl end of both AA and LA are identical up to ω -9, it is reasonable to assume that any discrimination between the two substrates would occur on the carboxylate end of the fatty acids, which is longer and more unsaturated for AA than LA. Therefore, one could hypothesize that subtle changes in this region of the protein structure, such as changing the volume of the active site or changing the charge interaction with the carboxylate on the fatty acid, such as R402 in 15-hLO-1, could affect the substrate specificity. With this hypothesis in mind, we compared the active site of the closed structure (inhibitor bound) to the open structure (no inhibitor) for 15-hLO-1 and observed 16 residues within 6 Å of the carboxylate end of the fatty acid that are different between the two structures (open and closed) (Figure 4 and Table 4). These findings indicate structural mobility in 15-hLO-1 and identify residues that could potentially affect substrate specificity. Interestingly, R402 does not move significantly between the open and closed structures, suggesting it is not involved in substrate specificity. This conclusion is supported by the fact that the 15-hLO-1(R402L) mutant does not elicit a change in the $(k_{\text{cat}}/K_m)^{\text{AA}}/(k_{\text{cat}}/K_m)^{\text{LA}}$ ratio compared to that of 15-hLO-1(WT) (17).

The analysis of the 15-hLO-2 is more complex. The active site cavities of the two 15-hLO isozymes are similar in shape, despite the fact that the two isozymes are only 55% similar in sequence (37% identical) and that most of the conserved residues discussed above for 15-hLO-1 are not present in the 15-hLO-2 cavity (Figure 5). Of the key residues suspected of substrate interaction for 15-hLO-1, only F365 is present in 15-hLO-2 (F352 in 15-hLO-1); there is no charged residue in the active site of 15-hLO-2 that could substitute for R402 in 15-hLO-1, nor is there a positively charged amino acid within 10 Å of the docked substrates (data not shown). These results suggest that substrate binding and recognition for 15-hLO-2 are fundamentally different from that of 15-hLO-1. This could explain the difference in their substrate specificity and their different responses to the addition of 13-HPODE, with 15-hLO-1 becoming more reactive against AA and 15-hLO-2 becoming less reactive. We are currently investigating the mechanism of substrate binding for both 15-hLO-1 and 15-hLO-2 using site-directed mutagenesis of the residues listed in Table 4, in the hopes of understanding substrate recognition more thoroughly and how it can be modified by allosteric regulation.

With respect to the cancer biology relevance of this discovery, it is possible that the product regulation of 15-hLO substrate specificity may have a direct effect on prostate cancer progression. As mentioned in the introductory section, 13-HPODE and 15-HPETE have opposing effects on cell proliferation, so a change in the substrate specificity of a

particular 15-hLO isozyme could initiate a change in the 13-HPODE:15-HPETE ratio in the cell and affect the cellular response. This hypothesis is consistent with the half-saturation points of 12-HPETE, 12-HETE, 13-HPODE, and 13-HODE (approximately 1 μM for each LO product) on 15-hLO-1 substrate specificity, since they are within the LO product concentration range seen in cells (from 0.6 to 2.0 μM) (58). The magnitude of the change in substrate specificity for the 15-hLO isozymes (3–5-fold) is also in the range seen in biology, 2–10-fold for RNR (59), suggesting that the allosteric effect in both 15-hLO-1 and 15-hLO-2 is feasible in the cell. We are currently investigating if LO products regulate substrate specificity in the cell and if there is any effect on cancer progression.

In conclusion, the key finding of this report is that LO products, and their reduced counterparts, bind to an allosteric site in the 15-hLO isozymes, affecting their substrate specificity. Considering that the allosteric effector molecule is an LO product itself, these *in vitro* results suggest that the 15-hLO isozymes manifest a product-feedback loop which could affect cellular biology and possibly be targeted for allosteric inhibition against cancer progression.

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